

**Program/Goal**

The program of this lab is to demonstrate the relative and absolute stability of various carbocation species and to experimentally determine the relative stability of various carbocation species and to determine the relative stability of various carbocation species.

**Introduction**

**Part I**

- 1. Prepare the three carbocation species:  $tert$ -butyl, isopropyl, and ethyl.
- 2. Add a small amount of concentrated hydrochloric acid to the reaction mixture.
- 3. The most reactive carbocation species is the  $tert$ -butyl cation. The next most reactive is the isopropyl cation. The least reactive is the ethyl cation.
- 4. Measure the relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol.
- 5. Compare the relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol with the relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol.
- 6. Measure the relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol.
- 7. Measure the relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol.

**Part II**

- 1. Add the relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol to the relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol.
- 2. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 3. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 4. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 5. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 6. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 7. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 8. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.

**Part III**

- 1. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 2. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 3. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 4. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 5. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
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- 7. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.
- 8. The relative rates of reaction of the three carbocation species with a small amount of  $tert$ -butyl alcohol are:  $tert$ -butyl > isopropyl > ethyl.

# Pkg Acp Che 112 Lab

**SA Dillow**



**Pkg Acp Che 112 Lab:**

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