



# Noise From The Underground

**L Cohen**



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ch 4 2 o 2 1 co 2 2 h 2 o 8 1 c 3 h 8 5 o 2 3 co 2 4

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1 1 n2 3 f2 2 nf3 2 2 c6h10 17 o2 12 co2 10 h2o 3 4 5 1 hbr 1 khco3 1 h2o 1 kbr 1 co2 2 gabr 3 3 na2so3 1 ga2 so3 3 6 nabr 3 sno 2 nf3 3 snf2 1 n2o3 using the following equation  
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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web may 28 2020 write the balanced equation for the reaction and calculate how much  $\text{sio}_2$  is required to produce 3.00 kg of  $\text{sic}$  answer  $\text{ce sio}_2 \text{ 3c} \rightarrow \text{sic 2co}$

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secant a ab b

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answer  $x^2 + y^2 = 9$  explanation  $r^2 = x^2 + y^2 = 4^2 + 3^2 = 25$   $r = 5$   $x^2 + y^2 = 25$  question 11

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not drawn to scale