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# Numbers Kiddy Bigs

**Raffaela Di Napoli**



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ch<sub>2</sub> ch ch<sub>2</sub> hc o h ci Ö o h<sub>11000</sub> h h b which are polar molecules c<sub>1</sub> h ch<sub>2</sub> ch<sub>2</sub> ch<sub>3</sub> sch ch<sub>2</sub> 1111100 C h h h Ö o o h ci Ö Ö d  
11 ci previous question next question

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determine based on electronegativity and symmetry if the following molecules are polar or nonpolar hcn h c n cf<sub>4</sub> c f<sub>4</sub> cocl<sub>2</sub>  
c o c l<sub>2</sub> nf<sub>3</sub> n f<sub>3</sub> q3

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as  $\text{BF}_3$  trigonal planar  $\text{CH}_4$  tetrahedral  $\text{PF}_5$  trigonal bipyramidal and  $\text{SF}_6$

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