

Standard Thermodynamic Values for Selected Substances*

Substance or ion	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/mol-K)	Substance or ion	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/mol-K)
e^- (g)	0	0	20.87	$\text{CaCO}_3(s)$	-1206.9	-1128.8	92.9
Aluminum				$\text{CaO}(s)$	-635.1	-603.5	38.2
$\text{Al}(s)$	0	0	28.3	$\text{Ca}(\text{OH})_2(s)$	-986.09	-898.56	83.39
$\text{Al}^{3+}(aq)$	-524.7	-481.2	-313	$\text{Ca}_3(\text{PO}_4)_2(s)$	-4158	-3899	263
$\text{AlCl}_3(s)$	-704.2	-628.9	110.7	$\text{CaSO}_4(s)$	-1432.7	-1320.3	107
$\text{Al}_2\text{O}_3(s)$	-1676	-1582	50.94	Carbon			
Barium				$\text{C}(\text{graphite})$	0	0	5.686
$\text{Ba}(s)$	0	0	62.5	$\text{C}(\text{diamond})$	1.896	2.866	2.439
$\text{Ba}(g)$	175.6	144.8	170.28	$\text{C}(g)$	715.0	669.6	158.0
$\text{Ba}^{2+}(g)$	1649.9	—	—	$\text{CO}(g)$	-110.5	-137.2	197.5
$\text{Ba}^{2+}(aq)$	-538.36	-560.7	13	$\text{CO}_2(g)$	-393.5	-394.4	213.7
$\text{BaCl}_2(s)$	-806.06	-810.9	126	$\text{CO}_2(aq)$	-412.9	-386.2	121
$\text{BaCO}_3(s)$	-1219	-1139	112	$\text{CO}_3^{2-}(aq)$	-676.26	-528.10	-53.1
$\text{BaO}(s)$	-548.1	-520.4	72.07	$\text{HCO}_3^-(aq)$	-691.11	-587.06	95.0
$\text{BaSO}_4(s)$	-1465	-1353	132	$\text{H}_2\text{CO}_3(aq)$	-698.7	-623.42	191
Boron				$\text{CH}_4(g)$	-74.87	-50.81	186.1
$\text{B}(\beta\text{-rhombohedral})$	0	0	5.87	$\text{C}_2\text{H}_2(g)$	227	209	200.85
$\text{BF}_3(g)$	-1137.0	-1120.3	254.0	$\text{C}_2\text{H}_4(g)$	52.47	68.36	219.22
$\text{BCl}_3(g)$	-403.8	-388.7	290.0	$\text{C}_2\text{H}_6(g)$	-84.667	-32.89	229.5
$\text{B}_2\text{H}_6(g)$	35	86.6	232.0	$\text{C}_3\text{H}_8(g)$	-105	-24.5	269.9
$\text{B}_2\text{O}_3(s)$	-1272	-1193	53.8	$\text{C}_4\text{H}_{10}(g)$	-126	-16.7	310
$\text{H}_3\text{BO}_3(s)$	-1094.3	-969.01	88.83	$\text{C}_6\text{H}_6(l)$	49.0	124.5	172.8
Bromine				$\text{CH}_3\text{OH}(g)$	-201.2	-161.9	238
$\text{Br}_2(l)$	0	0	152.23	$\text{CH}_3\text{OH}(l)$	-238.6	-166.2	127
$\text{Br}_2(g)$	30.91	3.13	245.38	$\text{HCHO}(g)$	-116	-110	219
$\text{Br}(g)$	111.9	82.40	174.90	$\text{HCOO}^-(aq)$	-410	-335	91.6
$\text{Br}^-(g)$	-218.9	—	—	$\text{HCOOH}(l)$	-409	-346	129.0
$\text{Br}^-(aq)$	-120.9	-102.82	80.71	$\text{HCOOH}(aq)$	-410	-356	164
$\text{HBr}(g)$	-36.3	-53.5	198.59	$\text{C}_2\text{H}_5\text{OH}(g)$	-235.1	-168.6	282.6
Cadmium				$\text{C}_2\text{H}_5\text{OH}(l)$	-277.63	-174.8	161
$\text{Cd}(s)$	0	0	51.5	$\text{CH}_3\text{CHO}(g)$	-166	-133.7	266
$\text{Cd}(g)$	112.8	78.20	167.64	$\text{CH}_3\text{COOH}(l)$	-487.0	-392	160
$\text{Cd}^{2+}(aq)$	-72.38	-77.74	-61.1	$\text{C}_6\text{H}_{12}\text{O}_6(s)$	-1273.3	-910.56	212.1
$\text{CdS}(s)$	-144	-141	71	$\text{C}_{12}\text{H}_{22}\text{O}_{11}(s)$	-2221.7	-1544.3	360.24
Calcium				$\text{CN}^-(aq)$	151	166	118
$\text{Ca}(s)$	0	0	41.6	$\text{HCN}(g)$	135	125	201.7
$\text{Ca}(g)$	192.6	158.9	154.78	$\text{HCN}(l)$	105	121	112.8
$\text{Ca}^{2+}(g)$	1934.1	—	—	$\text{HCN}(aq)$	105	112	129
$\text{Ca}^{2+}(aq)$	-542.96	-553.04	-55.2	$\text{CS}_2(g)$	117	66.9	237.79
$\text{CaF}_2(s)$	-1215	-1162	68.87	$\text{CS}_2(l)$	87.9	63.6	151.0
$\text{CaCl}_2(s)$	-795.0	-750.2	114	$\text{CH}_2\text{Cl}_2(g)$	-83.7	-60.2	234
				$\text{CH}_2\text{Cl}_2(l)$	-117	-63.2	179

*All values at 298 K.

(continued)

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