

# RADIOACTIVITY & NUCLEAR PHYSICS

## Structure of an atom

The atom consists of subatomic particles called electrons, protons and neutrons. The protons and neutron exist in the nucleus of the atom whereas the electrons orbit around the shell.



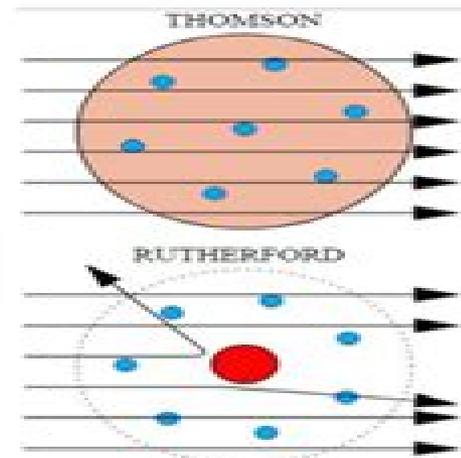
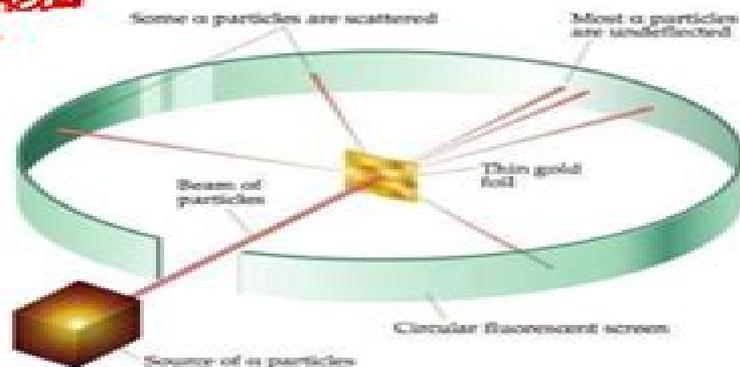
The atomic number tells the number of protons in an atom. The mass number tells the total number of protons and neutrons in an atom. So if the no. of neutrons are to be determined, we use the following formula.

$$\text{no. of neutrons} = \text{mass number} - \text{atomic number}$$

In a neutral atom, the no. of protons are equal to the no. of electrons.

## Geiger-Marsden Gold Foil Experiment

Alpha particles



At the start, it was assumed the atom consists of a large positive sphere with electrons stick on it. This was called the "plum pudding model". The experiment was to test the validity of the model but what turned out was entirely opposite.

# Radioactivity Nuclear Physics 3ed

**F Rizvi**



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## **Table of Contents Radioactivity Nuclear Physics 3ed**

1. Understanding the eBook Radioactivity Nuclear Physics 3ed
  - The Rise of Digital Reading Radioactivity Nuclear Physics 3ed
  - Advantages of eBooks Over Traditional Books
2. Identifying Radioactivity Nuclear Physics 3ed
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Radioactivity Nuclear Physics 3ed
  - User-Friendly Interface
4. Exploring eBook Recommendations from Radioactivity Nuclear Physics 3ed
  - Personalized Recommendations
  - Radioactivity Nuclear Physics 3ed User Reviews and Ratings
  - Radioactivity Nuclear Physics 3ed and Bestseller Lists
5. Accessing Radioactivity Nuclear Physics 3ed Free and Paid eBooks

- Radioactivity Nuclear Physics 3ed Public Domain eBooks
- Radioactivity Nuclear Physics 3ed eBook Subscription Services
- Radioactivity Nuclear Physics 3ed Budget-Friendly Options
- 6. Navigating Radioactivity Nuclear Physics 3ed eBook Formats
  - ePub, PDF, MOBI, and More
  - Radioactivity Nuclear Physics 3ed Compatibility with Devices
  - Radioactivity Nuclear Physics 3ed Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Radioactivity Nuclear Physics 3ed
  - Highlighting and Note-Taking Radioactivity Nuclear Physics 3ed
  - Interactive Elements Radioactivity Nuclear Physics 3ed
- 8. Staying Engaged with Radioactivity Nuclear Physics 3ed
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Radioactivity Nuclear Physics 3ed
- 9. Balancing eBooks and Physical Books Radioactivity Nuclear Physics 3ed
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Radioactivity Nuclear Physics 3ed
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Radioactivity Nuclear Physics 3ed
  - Setting Reading Goals Radioactivity Nuclear Physics 3ed
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Radioactivity Nuclear Physics 3ed
  - Fact-Checking eBook Content of Radioactivity Nuclear Physics 3ed
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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