

# What decides the properties of a polymer?

- Stronger attractive forces between chains = stronger, less flexible polymer.
- Chains able to slide past each other = flexible polymer .
- In poly(ethene) attractive forces are weak instantaneous dipole - induced dipole, will it be flexible or not?
- Nylon has strong hydrogen bonds, why does this make it a strong fibre?

# Properties Structure Of Polymers

**Robert R. Redfield**



## **Properties Structure Of Polymers:**

## Reviewing **Properties Structure Of Polymers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Properties Structure Of Polymers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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