

Acids and Bases and the Measurement of pH

Objectives

- To learn how to use pH electrodes to measure pH values
- To investigate the relative solubilities of acids and bases by measuring their pH values
- To measure the hydrolysis of salts
- To measure the relative strengths of acids by measuring the pH of the solutions

Experimental Data

Table 1:

Type of Solution	Measured pH of Solution	pH of Solution after Dilution
0.1M HCl	1.0	1.3
0.01M HCl	2.0	2.3
0.1M NaOH	13.0	12.7
0.01M NaOH	12.0	11.7
0.1M CH ₃ COOH	2.9	3.4
0.1M NH ₃	11.1	10.6

Table 2:

0.1M HCl (pH 1.0)

Volume of 0.1M HCl	Volume of 0.1M NaOH	Volume of 0.1M NaCl	Volume of 0.1M H ₂ O
10.0 mL	10.0 mL	10.0 mL	10.0 mL

Table 3:

0.1M HCl (pH 1.0)

0.1M NaOH (pH 13.0) + 0.1M CH₃COOH (pH 2.9) + 0.1M NH₃ (pH 11.1)

Example 1: pH calculations

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Raffaella Di Napoli



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