



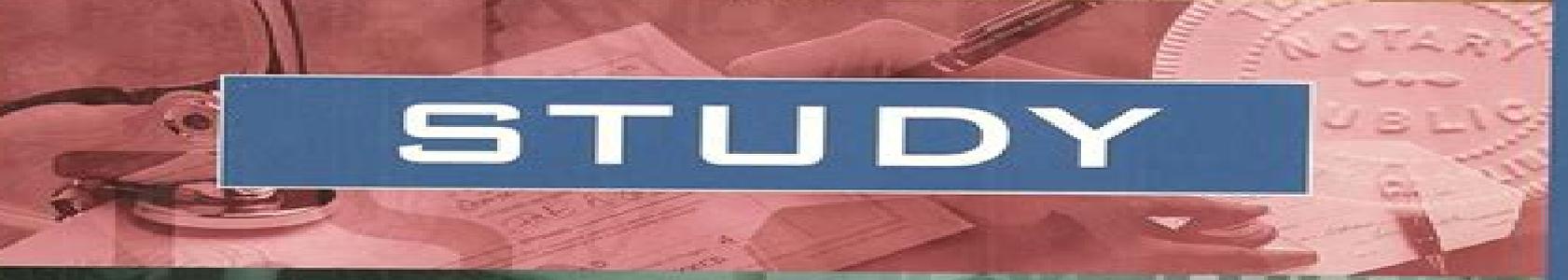
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David Kirk



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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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 $\text{a} + \text{i}_2 \rightarrow \text{o}_5$
 $5\text{co} + \text{i}_2 \rightarrow 5\text{co}_2$
 $\text{i}_2 + \text{o}_5 \rightarrow \text{i}_2\text{o}_5$
 $5\text{c} + \text{o}_2 \rightarrow 5\text{c} + \text{o}_2$

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 $2\text{sr} + \text{o}_2 \rightarrow 2\text{sro}$
 $\text{sr} \rightarrow \text{sr}^{2+}$ oxidized
 $\text{o}_2 \rightarrow \text{o}^{2-}$ reduced
 $\text{ox ag} = 2\text{li} + \text{s} \rightarrow \text{li}_2\text{s}$
 $\text{li} \rightarrow \text{li}^+$ oxidized
 $\text{red ag} = \text{s} \rightarrow \text{s}^{2-}$ reduced
 $\text{ox ag} = 3\text{2cs} + \text{br}_2$

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web here is some code to plot a cubic spline function

```
plot cubic spline x s0 s1 s2 s3 n length x inner points 20 for i 1 n 1 xx
linspace x i x i 1 inner points xi repmat x i 1 inner points yy s0 i s1 i xx xi s2 i xx xi 2 s3 i xx xi 3 plot xx yy b plot x i 0 r end
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```

web as expected it is zero to the left of 2 and rises like x^2 to the right of 2 plot xx subplot xx 2 3 y linewidth 3 axis 0 6 10 70 now we interpolate this particular cubic spline at the data sites 0 6 and plot the interpolant on top of the spline in black

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