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Module

# Operations on Polynomials

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# Operations On Polynomials Steps In Mathematics Modules

**Hussin A.Rothana**



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 $2S_3O_2 + 2SO_3 \rightarrow 3H_2O + 7914kJ$  is this endothermic or exothermic?

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 $2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$  energy endothermic the transfer of heat from the surroundings to the system  
 $2HgO(s) \rightarrow 2Hg(l) + O_2(g)$

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web  $q = 140\text{ g} \cdot 4.184\text{ J}/\text{g}\cdot\text{C} = 9372.16\text{ J} = 35.5\text{ g} \cdot x = 1\text{ mole KCN} \cdot 0.5453\text{ moles} = 65.1\text{ g} \cdot 0.5453\text{ mole} = 5$  when a 15 g sample of  $SO_2$  dissolves in 80 g of water in a calorimeter the temperature drops from 42 C to 15 C calculate the  $\Delta H$  for the process  
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