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Master Builders

Zhe Xu



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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells

filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is 1s 2 2s 2 2p 6 3s 2 3p 6 4s 1 or simply put at 4s 1 q2 what is the total number of electrons that can be filled in s p and d subshells

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