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vs

Reason

# Reason And Faith

**PT Brinkman**



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solvent usually a liquid and form a homogenous solution there are three factors that affect solubility solvent to determine whether a solute will dissolve in a solvent remember this saying like dissolves

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web when enough condensed water forms it flows back into the solution when this cooler water hits the bulk hot water it cools it down figure 2 2 2 2 2 2 c the solubility goes down and crystals form in a ring on the surface of the test tube at

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web conclusion for the expected results the purpose of performing these experiments was to achieve the true  $K_{sp}$  value of  $Ca(OH)_2$  which is 6.66 overall the combined experiment succeeded in showing that the results averaged between the

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