



# Savory Soups

**Jianjun Gao**



## **Savory Soups:**

## **Savory Soups** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Savory Soups**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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web Lewis structures represent the valence electrons present in the molecule as lone pairs unshared electrons drawn as dots or bonds shared electrons drawn as lines procedure for drawing Lewis structures add up the total number of valence electrons that each atom contributes to the molecule polyatomic ion

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web the PCl<sub>3</sub> Lewis structure has the typical case of phosphorus P in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that P and N generally have the same bonding structure note the lone pair dots without bonds on top of P just like for N in the previous example for NH<sub>3</sub> chlorine Cl

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diagrams that show the bonding between atoms of a molecule and the lone pairs of

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the  
periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and  
lone pairs surrounding the central atom and lastly identify the structure geometry